

This dataset contains:

Analytical characterisation of the new complexes (CHN microanalysis).

Thermogravimetric analyses.

X-ray Crystallographic data:

- Structure of **1Cu**·2H<sub>2</sub>O at 150 K (CCDC 1973396)
- Structure of **1Cu** phase B at 350 K (CCDC 1973398)
- Structure of **1Cu** phase B at 300 K (CCDC 1973393)
- Structure of **1Cu** phase C' at 250 K (CCDC 1973394)
- Structure of **1Cu** phase C' at 200 K (CCDC 1973395)
- Structure of **1Cu** phase C' at 150 K (CCDC 1973397)
- Structure of **1Zn**·2H<sub>2</sub>O at 150 K (CCDC 1973389)
- Structure of **1Zn** phase B at 305 K (CCDC 1973390)
- Structure of **1Zn** phase C at 285 K (CCDC 1973391)
- Structure of **1Zn** phase C at 240 K (CCDC 1973392)
- Structure of **1Zn** phase C at 150 K (CCDC 1973368).

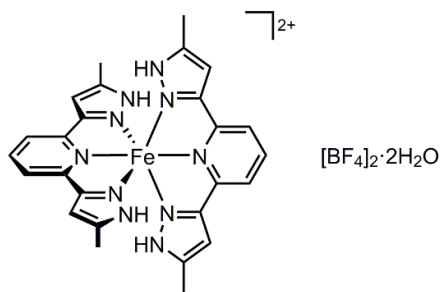
X-ray powder diffraction data collected with Cu- $K_{\alpha}$  and synchrotron radiation (measured and simulated).

Solid state magnetic susceptibility measurements (raw and processed data).

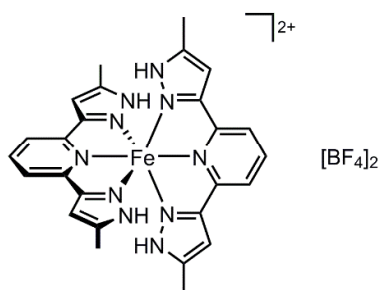
Differential Scanning Calorimetry (DSC) data.

X-band and Q-band EPR spectra (measured and simulated).

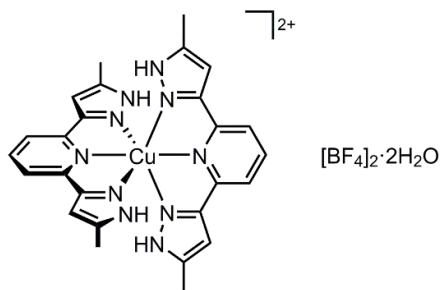
## Compounds prepared during this study



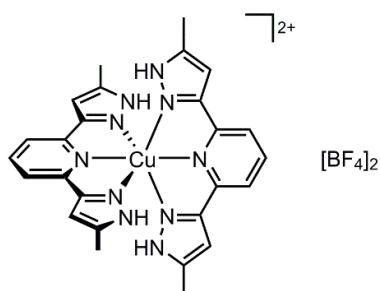
*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)iron(II) ditetrafluoroborate dihydrate  
 $[FeL_2][BF_4]_2 \cdot 2H_2O$   
 $C_{26}H_{30}B_2F_8FeN_{10}O_2$   
**1Fe**·2H<sub>2</sub>O



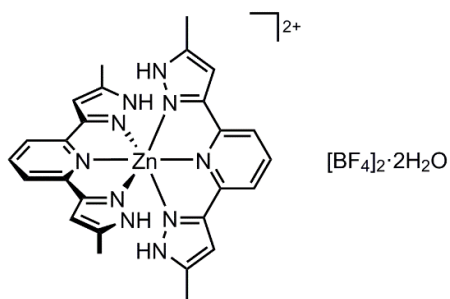
*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)iron(II) ditetrafluoroborate  
 $[FeL_2][BF_4]_2$   
 $C_{26}H_{26}B_2F_8FeN_{10}$   
**1Fe**



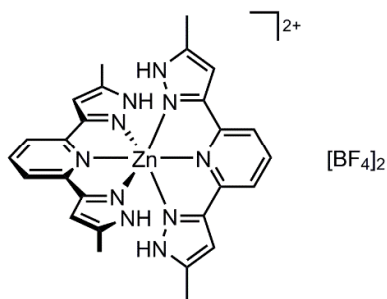
*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)copper(II) ditetrafluoroborate dihydrate  
 $[CuL_2][BF_4]_2 \cdot 2H_2O$   
 $C_{26}H_{30}B_2CuF_8N_{10}O_2$   
**1Cu**·2H<sub>2</sub>O



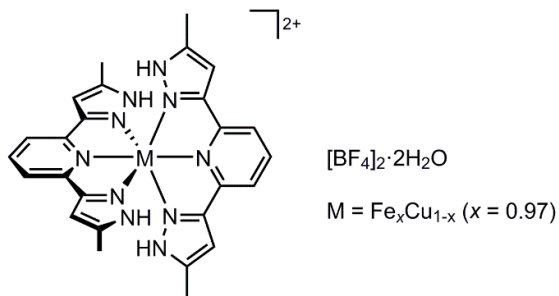
*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)copper(II) ditetrafluoroborate  
 $[CuL_2][BF_4]_2$   
 $C_{26}H_{26}B_2CuF_8N_{10}$   
**1Cu**



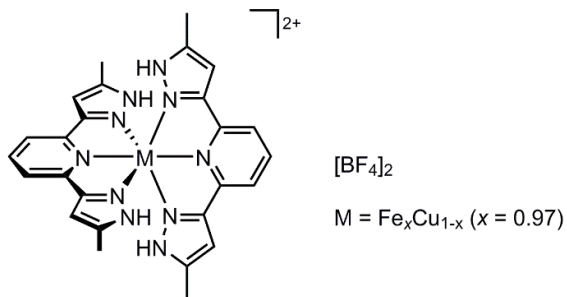
*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)zinc(II) ditetrafluoroborate dihydrate  
 [ZnL<sub>2</sub>][BF<sub>4</sub>]<sub>2</sub>·2H<sub>2</sub>O  
 C<sub>26</sub>H<sub>30</sub>B<sub>2</sub>F<sub>8</sub>N<sub>10</sub>O<sub>2</sub>Zn  
 1Zn·2H<sub>2</sub>O



*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)zinc(II) ditetrafluoroborate  
 [ZnL<sub>2</sub>][BF<sub>4</sub>]<sub>2</sub>  
 C<sub>26</sub>H<sub>26</sub>B<sub>2</sub>F<sub>8</sub>N<sub>10</sub>Zn  
 1Zn



*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)iron,copper(II) ditetrafluoroborate dihydrate solid solution  
 [Fe<sub>0.97</sub>Cu<sub>0.03</sub>L<sub>2</sub>][BF<sub>4</sub>]<sub>2</sub>·2H<sub>2</sub>O  
 C<sub>26</sub>H<sub>30</sub>B<sub>2</sub>Cu<sub>0.03</sub>F<sub>8</sub>Fe<sub>0.97</sub>N<sub>10</sub>O<sub>2</sub>  
 1Fe,Cu·2H<sub>2</sub>O



*Bis*-(2,6-*bis*[5-methyl-1*H*-pyrazol-3-yl]pyridine)iron,copper(II) ditetrafluoroborate solid solution  
 [Fe<sub>0.97</sub>Cu<sub>0.03</sub>L<sub>2</sub>][BF<sub>4</sub>]<sub>2</sub>  
 C<sub>26</sub>H<sub>26</sub>B<sub>2</sub>Cu<sub>0.03</sub>F<sub>8</sub>Fe<sub>0.97</sub>N<sub>10</sub>  
 1Fe,Cu